

An Enantioconvergent Route to (-)-Kainic Acid

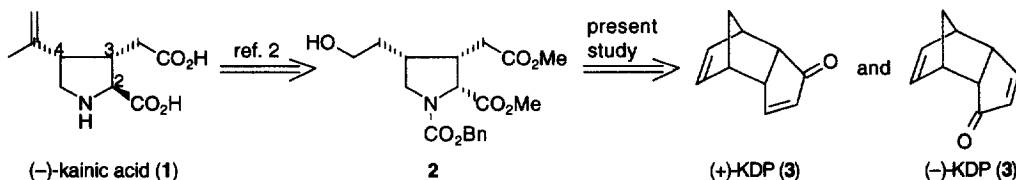
Yoshihisa Nakada, Tsutomu Sugahara, and Kunio Ogasawara*

Pharmaceutical Institute, Tohoku University, Aobayama, Sendai 980-77, Japan

Fax +81-22-217-6845; E-mail c21799@cct.u.cc.tohoku.ac.jp

Abstract: A new stereocontrolled route to (-)-kainic acid, the representative of the kainoid amino acids and exhibiting insecticidal, anthelmintic and neuroexcitatory properties, has been devised in an enantioconvergent way using either (+)- or (-)-ketodicyclopentadiene as the starting material by employing a concurrent retro-Diels-Alder reaction and intramolecular ene reaction as the key step.
 © 1997, Elsevier Science Ltd. All rights reserved.

The kainoid amino acids have attracted considerable interest because of their pronounced insecticidal, anthelmintic and neuroexcitatory properties.¹ Among these properties, the neuroexcitatory activity is attributed to their *trans*-C-2/C-3:*cis*-C-3/C-4 structure and the functionality on the C-4 center besides 2-carboxy-3-carboxymethyl functionalities. Accordingly, an efficient enantiocontrolled procedure being capable of producing the appropriate 2,3,4-trisubstituted pyrrolidines carrying a versatile C-4 functionality has been sought. We report here a new stereocontrolled route to (-)-kainic acid (1) in a formal sense by the synthesis of the key intermediate² 2 carrying a versatile 4-(2-hydroxyethyl) group starting either from the (+)- or (-)-enantiomer of optically pure ketodicyclopentadiene^{3,4} (KDP) (3) by employing a concurrent one-pot retro-Diels-Alder reaction and intramolecular ene reaction as the key step (Scheme 1).

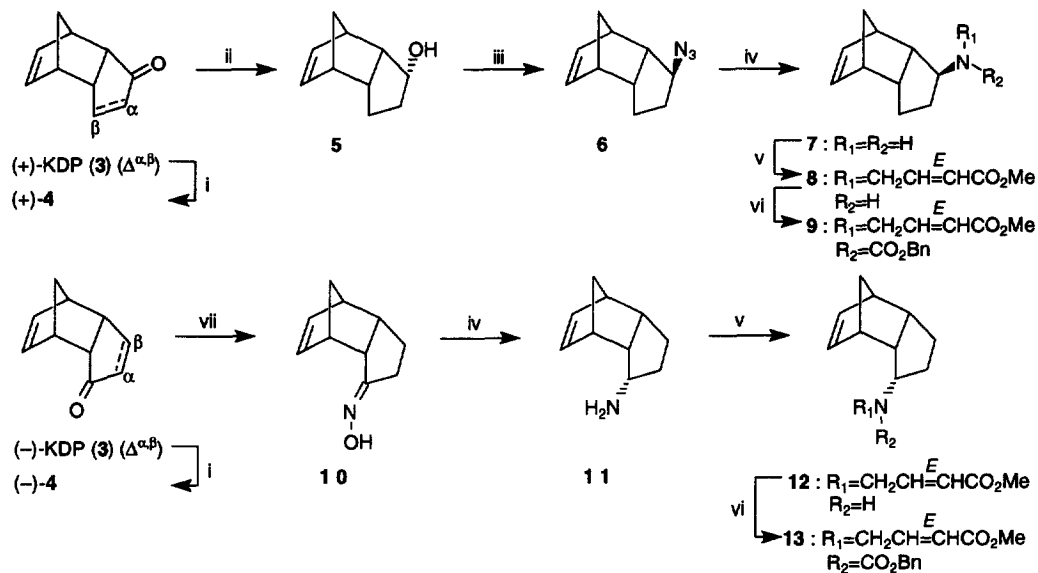


Scheme 1

(+)-KDP [(+)-3] (>99% ee) was transformed stereoselectively into the *endo*-alcohol⁵ 5, mp 96 °C, $[\alpha]_D^{28}$ -13.1 (c 0.5, CHCl₃), in 88% yield via the (+)-ketone⁶ (+)-4. The Mitsunobu reaction⁷ of 5 with diphenylphosphoryl azide (DPPA) gave the *exo*-azide 6, $[\alpha]_D^{27}$ +70.4 (c 1.0, CHCl₃), with inversion which was sequentially transformed into the *exo*-carbamate 9, $[\alpha]_D^{25}$ +49.1 (c 1.0, CHCl₃), having the *N*-(*E*-3-

ethoxycarbonyl-2-propenyl) group via the *exo*-primary amine **7** and the *exo*-secondary amine **8**. Overall yield of **9** from **5** was 44% in four steps.

On the other hand, (-)-KDP [(-)-**3**] (>99% ee) was transformed into the single oxime **10**, $[\alpha]_D^{27} -279.5$ (*c* 1.3, CHCl₃), via the (-)-ketone⁶ (-)-**4**. On sequential stereoselective reduction, *N*-alkylation, and carbamoylation, **10** furnished the *endo*-carbamate **13**, $[\alpha]_D^{27} -83.8$ (*c* 1.6, CHCl₃), via the *endo*-primary amine **11** and the *endo*-secondary amine **12**. Overall yield of **13** from (-)-**3** was 49% in five steps (Scheme 2).



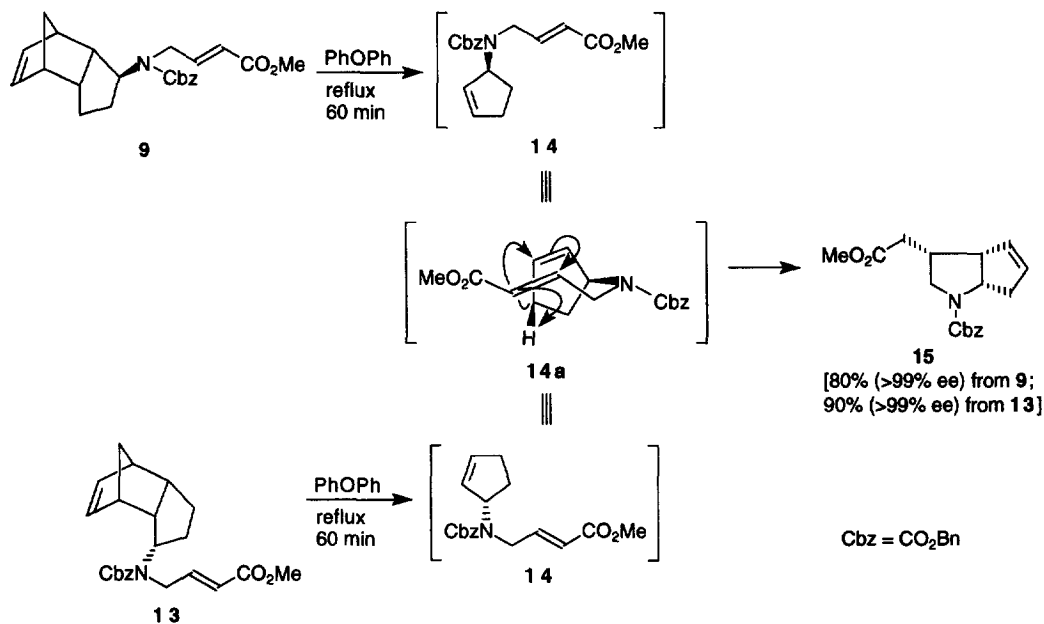
Scheme 2

Reagents and conditions: i, Zn, AcOH-EtOH (1:4), reflux [97% for (-)-**4**]; ii, NaBH₄, MeOH, -78 °C [88% from (+)-**3**]; iii, DPPA, (NCO₂Et)₂, PPh₃, THF, 0 °C ~ room temp. (94%); iv, LiAlH₄, THF, room temp.; v, BrCH₂CH=CHCO₂Me, Et₃N, DMF, 0 °C; vi, ClCO₂Bn, NaH, DMF, -20 °C [47% for **9** from **6** and 49% for **13** from (-)-**3**]; vii, NH₂OH·HCl, pyridine, room temp. (97%).

Having obtained the *exo*-**9** and the *endo*-**13** diastereomers, we next examined the thermolysis of these compounds in hot diphenyl ether. Since we had learned that the retro-Diels-Alder reaction of certain substrates having a bicyclo[2.2.1]heptene system relating to **9** and **13** was best carried out in diphenyl ether at boiling point⁸ (~260 °C), we expected both of these would form the same monocyclic carbamate **14** having a 1,6-diene system which further would undergo the intramolecular ene reaction⁹ under the same conditions to give the bicyclo product **15** having a trisubstituted pyrrolidine framework.

When the *exo*-**9** was heated in diphenyl ether at refluxing temperature, the reaction terminated within 60 min to give rise to the bicyclic product **15**, $[\alpha]_D^{27} +95.8$ (*c* 1.0, CHCl₃), in 80% yield as the single isomer as expected. On the same treatment, the *endo*-diastereomer **13** afforded the same bicyclic product **15**, $[\alpha]_D^{30} +94.7$ (*c* 1.1, CHCl₃), in 90% yield as a single isomer. The optical purities of the product **15** from both **9** and **13** were determined to be >99% ee by hplc using a chiral column (CHIRALCEL OD, elution with 5% *i*-

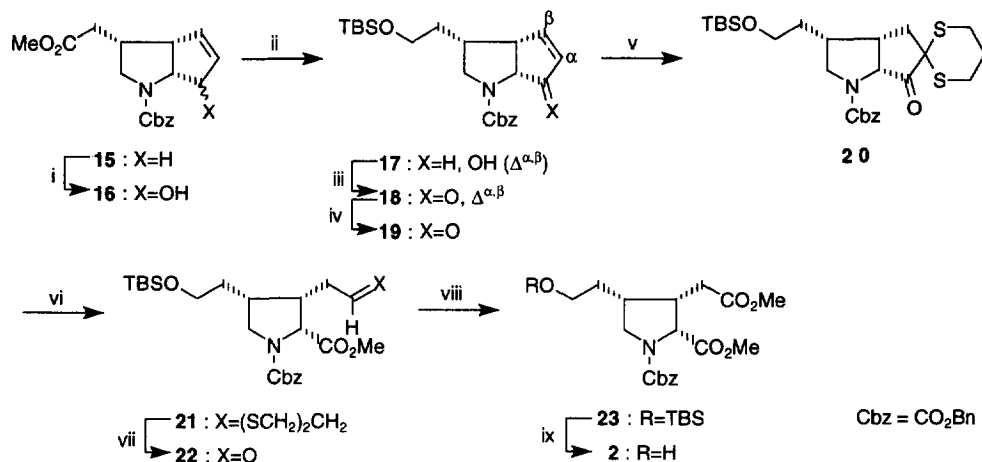
PrOH/hexane), indicating that the original chiral integrity of the precursor carbamates was preserved under these thermal conditions. The relative stereochemistry of **15** was determined to be the all-*cis* configuration as shown by facile iodolactone formation (single product in 73% yield) when the carboxylate generated from **15** by saponification was treated with iodine in acetonitrile¹⁰; its absolute stereochemistry, however, was determined after converting it into the key intermediate **2** (see below). These findings led us to conclude that the intramolecular ene reaction of the transient 1,6-diene **14** took place in a stereospecific manner taking the most unstrained transition state **14a** to afford the all-*cis* product **15** (Scheme 3).



Scheme 3

In order to obtain the key intermediate **2** of (–)-kainic acid (**1**), the ene product **15** was first treated with selenium(IV) oxide in the presence of formic acid¹¹ to give the allylic alcohol **16** in 68% yield as a diastereomeric mixture. On sequential hydride reduction, selective *O*-silylation, manganese(IV) dioxide oxidation, and the cuprate-mediated 1,4-reduction,¹² **16** furnished the bicyclic ketone **19**, $[\alpha]_D^{27} +106.6$ (*c* 1.3, CHCl₃), in 58% overall yield *via* the silyl ether **17** and the enone **18**, $[\alpha]_D^{27} +124.0$ (*c* 1.6, CHCl₃). The ketone **19** was transformed into the α -diketone monothioacetal¹³ **20**, $[\alpha]_D^{21} -21.8$ (*c* 1.1, CHCl₃), in 85% yield which was cleaved under basic conditions¹³ to give the dithiane **21**, $[\alpha]_D^{28} -7.1$ (*c* 1.0, CHCl₃), in 60% yield after esterification. Hydrolysis of the dithiane group¹³ of **21** afforded the aldehyde **22**, $[\alpha]_D^{27} +19.2$ (*c* 0.7, CHCl₃), in 90% yield which, on oxidation with a mixture of iodine and potassium hydroxide in methanol,¹⁴ furnished the key intermediate **2**, $[\alpha]_D^{31} +13.1$ (*c* 0.3, CHCl₃) [lit.^{2b}: $[\alpha]_D^{27} +11.7$ (*c* 0.5, CHCl₃)], in 61% yield after desilylation¹⁵ of the resulting monocyclic silyl ether **23**, $[\alpha]_D^{28} +32.8$ (*c* 0.1, CHCl₃) (Scheme 4).

As we have obtained^{2a} (–)-kainic acid (**1**) from **2**, the present synthesis constitutes a formal enantio-convergent synthesis starting from both (+)- and (–)-enantiomers of KDP (**3**). However, the present synthesis



Scheme 4

Reagents and conditions: i, SeO₂ (2 equiv.), HCO₂H (2 equiv.), dioxane, 90 °C, 4 h (68%); ii, LiAlH₄, THF (87%) then *t*-BuMe₂SiCl, Et₃N, CH₂Cl₂, room temp. (89%); iii, MnO₂, benzene-CH₂Cl₂ (1:1), room temp. (81%); iv, *i*-Bu₂AlH, CuI, HMPA-THF (1:4), -78 °C (92%); v, pyrrolidine, benzene, reflux then (TsSCH₂)₂CH₂, Et₃N, MeCN, reflux (85%); vi, *t*-BuOK (1.5 equiv.), *t*-BuOH, trace H₂O, -50 °C, acid workup then CH₂N₂ (60%); vii, MeI, CaCO₃, H₂O-MeCN (1:4), reflux (90%); viii, I₂, KOH, MeOH, -10 °C; ix, HF-pyridine, THF, room temp. (61% from 22).

has more potential value for the stereocontrolled construction of a variety of modified kainoid amino acids for biological investigation based on the functionality of the tricyclic starting material **3** which allows various stereocontrolled modifications⁴ not only on the side chain moieties, but also on the core pyrrolidine framework.

REFERENCES AND NOTES

- For pertinent reviews on the kainoid amino acid chemistry, see: (a) Hashimoto, K.; Shirahama, H. *J. Syn. Org. Chem. Jpn.* **1989**, *47*, 212-223. (b) Hashimoto, K.; Shirahama, H. *Trends Org. Chem.* **1991**, *2*, 1-32. (c) Parsons, A. F. *Tetrahedron* **1996**, *52*, 4149-4174.
- (a) Takano, S.; Iwabuchi, Y.; Ogasawara, K. *J. Chem. Soc., Chem. Commun.* **1988**, 1204-1206. (b) Kawamura, M.; Ogasawara, K. *Heterocycles* **1997**, *44*, 129-132.
- For a recent synthesis of optically pure KDP, see: Sugahara, T.; Kuroyanagi, Y.; Ogasawara, K. *Synthesis* **1996**, 1101-1108.
- For a pertinent review, see: Ogasawara, K. *J. Syn. Org. Chem. Jpn.* **1996**, *54*, 15-26.
- Satisfactory spectral [ir, ¹H nmr (300 MHz), and mass] and analytical (combustion and/or high resolution mass spectrometric) data were obtained for all new compounds isolated.
- Takano, S.; Sato, T.; Inomata, K.; Ogasawara, K. *J. Chem. Soc., Chem. Commun.* **1991**, 462-464.
- Lal, B.; Pramanik, B. N.; Manhas, M. S.; Bose, A. K. *Tetrahedron Lett.* **1977**, 1977-1980.
- Takano, S.; Kamikubo, T.; Moriya, M.; Ogasawara, K. *Synthesis* **1994**, 601-604.
- For a pertinent review, see: Oppolzer, W.; Snieckus, V. *Angew. Chem. Int. Ed. Engl.* **1978**, *17*, 476-486.
- For a pertinent review, see: Dowle, M. D.; Davies, D. I. *Chem. Soc. Rev.* **1979**, 171-197.
- Shibuya, K. *Synth. Commun.* **1994**, *24*, 2923-2941.
- Tsuda, T.; Fujii, T.; Kawasaki, K.; Saegusa, T. *J. Chem. Soc., Chem. Commun.* **1980**, 1013-1014.
- For a pertinent review, see: Takano, S.; Ogasawara, K. *J. Syn. Org. Chem. Jpn.* **1977**, *35*, 795-813.
- Yamada, S.; Morizono, D.; Yamamoto, K. *Tetrahedron Lett.* **1992**, *33*, 4329-4332.
- Nicolaou, K. C.; Webber, S. E. *Synthesis* **1986**, 453-461.

(Received in Japan 21 November 1996; revised 12 December 1996; accepted 16 December 1996)